Highly Sensitive Bulk Silicon Chemical Sensors with Sub-5 nm Thin Charge Inversion Layers

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ABSTRACT: There is an increasing demand for mass-producible, low-power gas sensors in a wide variety of industrial and consumer applications. Here, we report chemical-sensitive field-effect-transistors (CS-FETs) based on bulk silicon wafers, wherein an electrostatically confined sub-5 nm thin charge inversion layer is modulated by chemical exposure to achieve a high-sensitivity gas-sensing platform. Using hydrogen sensing as a “litmus” test, we demonstrate large sensor responses (>1000%) to 0.5% H₂ gas, with fast response (<60 s) and recovery times (<120 s) at room temperature and low power (<50 μW). On the basis of these performance metrics as well as standardized benchmarking, we show that bulk silicon CS-FETs offer similar or better sensing performance compared to emerging nanostructures (few nanometers thin metal alloys) sensitive to different gases. While such nanoscale silicon can provide high sensitivity, thickness uniformity control across wafers can lead to process, cost, and yield complexities in large-scale manufacturing. On the other hand, conventional bulk silicon transistors do not have a physically thin channel, resulting in less susceptibility to such complexities, and can be manufactured very economically.

In this work, we demonstrate bulk silicon CS-FETs as a highly sensitive low-power gas-sensing platform. The well-established concept of few nanometers thin inversion layers in conventional MOSFETs is adopted here using proper device architecture and operating voltage conditions. Through the electrostatic confinement of the inversion layer, the capacitive coupling between the sensing layer and channel is maximized, enabling high detection sensitivity. To evaluate this platform, hydrogen gas sensing is used as the test application. Monitoring hydrogen leaks is becoming increasingly important in several industrial and consumer applications. Here, we report chemical-sensitive field-effect-transistors (CS-FETs) based on bulk silicon wafers, wherein an electrostatically confined sub-5 nm thin charge inversion layer is modulated by chemical exposure to achieve a high-sensitivity gas-sensing platform. Using hydrogen sensing as a “litmus” test, we demonstrate large sensor responses (>1000%) to 0.5% H₂ gas, with fast response (<60 s) and recovery times (<120 s) at room temperature and low power (<50 μW). On the basis of these performance metrics as well as standardized benchmarking, we show that bulk silicon CS-FETs offer similar or better sensing performance compared to emerging nanostructures (few nanometers thin metal alloys) sensitive to different gases. While such nanoscale silicon can provide high sensitivity, thickness uniformity control across wafers can lead to process, cost, and yield complexities in large-scale manufacturing. On the other hand, conventional bulk silicon transistors do not have a physically thin channel, resulting in less susceptibility to such complexities, and can be manufactured very economically.

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applications, requiring stable sensors that can detect below the lower explosion limit of 4% (v/v in air) at low power, low cost and with a very small form factor.26

**DEVICE DESCRIPTION**

Conceptually, bulk silicon CS-FETs are similar to conventional enhancement-mode silicon transistors with the exception of the electrically active gate stack that is replaced by a large surface area, ultrathin chemical sensing layer, as depicted in Figure 1.

![Figure 1. Cross-sectional schematic of a bulk silicon CS-FET with electrostatic confinement of the charge inversion layer for achieving high sensitivity.](image)

This sensing layer is electrically floating and can be engineered to be sensitive to a target gas, where interactions can lead to reversible changes in work function and/or morphology. For the purpose of this work, ultrathin sensing layers composed of Ni (0.3 nm) and Pd (1 nm) are used for H2 gas sensing, where H2 readily dissociates over Pd at room temperature into atomic hydrogen, leading to the formation of PdHx.27 CS-FETs are readily available over Pd at room temperature into atomic hydrogen, leading to the formation of PdHx.27 CS-FETs are configured as n-type transistors with light p-body doping (∼8 × 1014 boron atoms cm−2), and the electrically floating sensing layer is capacitively coupled to the silicon channel via the native oxide (effective oxide thickness, EOT, of 2.5 to 3 nm). The sensitivities of these sensors are dependent on the threshold voltage of the transistors.

Under equilibrium and ambient conditions, the CS-FET threshold voltage (VT) is determined by the body doping and the effective work function (EWF) of the sensing layer, which for ultrathin Ni–Pd is expected to be lower (∼4.2 eV) than bulk values (∼5.11 eV) due to work-function dependence on metal thickness.28 If the VT is sufficiently low, an inversion layer of electrons (transistor channel) is created at the Si/SiO2 interface. The total electron density and thickness of the inversion layer (which is directly dependent on VT) can then be controlled by applying a reverse bias to the silicon substrate (also called body), as depicted in Figure 1. In conventional transistors, this mechanism of VT control is called the “body effect”, where an appropriate body voltage (VSUB) effectively controls the p–n junction formed between the p-body and the n-inversion layer. From a sensors perspective, this provides a highly tunable operation for CS-FETs, where the device can be tuned to the optimal performance by using VSUB. Equation 1 describes the relation between applied body bias and threshold voltage:

\[
\Delta V_T = \sqrt{2\varepsilon_s q N_{SUB} \left( (2\Phi_F + (V_T - V_{SUB})^{1/2} - (2\Phi_F)^{1/2} \right)}
\]

where εs is the dielectric permittivity of silicon, q is the electron charge, NSUB is the body doping, Cox is the capacitance of the native oxide, VT is the source voltage (ground), ΦF is the potential difference between the mid-gap and Fermi energy levels of the silicon body and VSUB is the applied body bias.

**RESULTS AND DISCUSSION**

**Device Modeling and Simulation.** A physical understanding of the sensing mechanism in bulk silicon CS-FET gas sensors is presented in Figure 2 using TCAD (Synopsys v:2016) device modeling and simulations. Table 1 lists the parameters used in simulating the n-type transistors where the sensing layer work function is set at 4.2 eV. Details on device modeling and simulation are in the Methods section.

![Figure 2. Characteristic of simulated silicon CS-FET devices depicting (a) inversion layer profiles, (b) extracted inversion layer thickness (sensing layer work function (Φm) is set to 4.2 eV), and (c) peak electron density at different body biases. (Note: Inversion layer profiles are extracted across the channel midpoint of the simulated device.)](image)

Table 1. CS-FET Device Parameters

<table>
<thead>
<tr>
<th>parameter</th>
<th>value</th>
</tr>
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<tbody>
<tr>
<td>gate length (Lg)</td>
<td>3 μm</td>
</tr>
<tr>
<td>effective oxide thickness (EOT)</td>
<td>3 nm</td>
</tr>
<tr>
<td>source/drain doping</td>
<td>1 × 1020 cm−3, phosphorus</td>
</tr>
<tr>
<td>body doping (NSUB)</td>
<td>8 × 1020 cm−3, boron</td>
</tr>
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As it can be seen in Figure 2a, at VSUB = 0 V, a simulated peak electron density of 2 × 1016 cm−3 is observed at an inversion layer depth of 3.3 nm with a total inversion layer thickness (Tinv) of 17.4 nm. Tinv is extracted as the location of the charge centroid in Figure 2a, and the electron density distribution is extracted across the midpoint of the CS-FET silicon channel. Applying VSUB = −4 V lowers the peak electron density to 0.65 × 1016 cm−3 at an inversion layer depth of 2.7 nm and a Tinv of 6.3 nm. With higher reverse body bias, the inversion layer is not only thinner but also pushed closer to the interface of the native oxide.
oxide and silicon, leading to improved electrostatic control in the channel by the sensing layer. As will be seen in the subsequent sections, this drastically improves sensor response and sensitivity.

Next, we explore various device parameters that can be varied to optimize the charge inversion layer. $T_{\text{inv}}$ is inversely dependent on body doping ($N_{\text{SUB}}$), as shown in Figure 2b. As $N_{\text{SUB}}$ is increased from $8 \times 10^{14}$ cm$^{-3}$ to $8 \times 10^{16}$ cm$^{-3}$ of boron atoms, $T_{\text{inv}}$ (at $V_{\text{SUB}} = -4$ V, EOT = 3 nm) is reduced from 6.3 nm to 2 nm. Body doping and charge density in the inversion layer are also inversely related to each other. As indicated in Figure 2c, increasing $N_{\text{SUB}}$ from $8 \times 10^{14}$ cm$^{-3}$ to $8 \times 10^{16}$ cm$^{-3}$ of boron, leads to a decrease in peak electron density ($n_{\text{electron}}$) from $0.65 \times 10^{16}$ cm$^{-3}$ to $0.4 \times 10^{11}$ cm$^{-3}$ ($V_{\text{SUB}} = -4$ V, EOT = 3 nm). The effect of oxide thickness on $T_{\text{inv}}$ is minimal, as seen in Figure 2b, where only at low $V_{\text{SUB}}$ does EOT reduction lead to minimal decrease in $T_{\text{inv}}$. Reducing the EOT from 5 nm to 1 nm leads to an increase in peak electron density from $0.5 \times 10^{15}$ cm$^{-3}$ to $1 \times 10^{14}$ cm$^{-3}$ ($V_{\text{SUB}} = -4$ V, $N_{\text{SUB}} = 8 \times 10^{16}$ cm$^{-3}$). Based on this discussion, body doping and effective oxide thickness are key optimization parameters to consider in designing a sensitive bulk CS-FET gas sensor. However, it is important to note that $T_{\text{inv}}$, $n_{\text{electron}}$, EOT, and $N_{\text{SUB}}$ are intricately linked to each other and that the above discussion provides highly generalized guidelines for CS-FET sensor design based on simulation results.

**Experimental Validation of Correlation between Sensitivity and Inversion Layer Thickness.** Bulk silicon CS-FETs are fabricated using a fully CMOS-compatible, gate-last processing scheme (see Methods section and Supporting Information S1), where the Ni–Pd sensing layer is deposited in the penultimate process step. Following this, the sensor is annealed in N$_2$ at 150 °C for 1 h. Figure 3a shows the experimentally measured room-temperature sensor response of a Ni–Pd CS-FET to different concentrations of hydrogen ranging from 0.05% to 0.5% (diluted in dry air) in steps of 0.05%, at different body biases. Details on the measurement apparatus can be found in the Methods section. With increasing reverse body bias from 0 V to $-2$ V, % sensor response (($I_{\text{peak}} - I_{\text{baseline}})/I_{\text{baseline}}$) to 0.5% H$_2$ concentration increases from 291% to 1383%, as indicated in Figure 3b. Furthermore, sensor linearity is also drastically improved, where the sensitivity increases from 0.04%/ppm to 0.27%/ppm upon changing body biases from 0 V to $-2$ V. Here, sensitivity is defined as the slope of the % sensor response (($I_{\text{peak}} - I_{\text{baseline}})/I_{\text{baseline}}$) per ppm of hydrogen gas. It is to be noted that variations in processing conditions of the sensing layer, for example, annealing in forming gas instead of N$_2$, can result in high sensor responses.
This is due to formation of large-size Pd clusters providing increased surface area for H₂ interaction. However, this negatively impacts sensor response time due to longer H₂ diffusion paths. It is also important to note that bare silicon CS-FETs without any sensing layers do not show any response to hydrogen (see Supporting Information S3). Figure 3c shows the sensor response time (t₉₀) vs hydrogen concentration, with minimum and maximum t₉₀ ≈ 36 s (for 0.5% H₂) and 196 s (for 0.05% H₂), respectively. t₉₀ is defined as the time taken for the sensor to reach 90% of its peak response value from the baseline current. Varying the body bias appears to have no significant effect on sensor response time. This is expected, as response times are dependent on the rate at which hydrogen diffuses and adsorbs on the Ni–Pd sensing layer. It is important to note that silicon CS-FETs will have a temperature dependence, requiring appropriate compensation for harsh environment operation. Results pertaining to this will be described in a future work. In all of the above measurements, the total power consumption of the hydrogen sensors is below 50 μW, reaffirming bulk silicon CS-FETs as a low-power gas-sensing platform.

Figure 4 compares the experimental data to simulation results, where simulated sensor responses are obtained at different body biases for a constant −0.1 eV work-function change in the sensing layer (analogous to a simulated gas exposure of 0.5% H₂). As shown, the trend of increasing sensor responses with higher reverse body bias is consistent in both theory and experiment. However, the simulated work-function change does not capture the interaction between hydrogen and the sensing layer, which may explain the discrepancy in the trends.

**Sensor Hysteresis, Ambient Drift, and Long-Term Stability.** Several experiments were performed to gauge sensor hysteresis and long-term stability. Exposing the sensor to cycles of low (0.05%), medium (0.25%), and high concentrations (0.5%) of hydrogen indicates minimal hysteresis in sensor performance, as indicated by Figure 5a. Figure 5b captures the baseline drift of two sensors (V_DS = 3 V, V_SUB = 0 V) over a period of 5.7 days, where the sensors were measured in ambient air without any gas flow and uncontrolled room humidity. The maximum variation from mean baseline current in both sensors is approximately 10%, indicating stable sensor baselines. Additionally, bare silicon CS-FETs without any sensing layers also exhibit similar stability as depicted in Supporting Information S5. Figure 5c shows the variation in peak sensor response current to a fixed hydrogen concentration for nearly a week, where a sensor is exposed to 0.5% H₂ (for 10 min) once per day. This measurement was done at room temperature with the relative humidity left uncontrolled and varying between 20% and 40%. Based on these results, the CS-FET platform exhibits minimal sensor performance degradation and long-term stability. With respect to sensor selectivity, we previously demonstrated the Ni–Pd sensing layer to be selective to H₂S (hydrogen sulfide) and NO₂ (nitrogen dioxide) via multiplexed gas-sensing gas experiments. Detailed selectivity results of the bulk silicon CS-FETs in contextually defined applications will be described in a future work.

**Benchmarks Comparison with Emerging Low-Dimensional Materials.** Finally, the performance of bulk silicon CS-FETs was benchmarked against emerging materials such as carbon nanotubes, MoS₂, and graphene for hydrogen gas sensing at the same concentration level (0.5%). Sensitive and fast detection at this concentration is important from a safety perspective, as it is below the lower explosion limit of 4%. This benchmark cites research works that have used both functionalized and nonfunctionalized materials. As indicated in Figure 6, bulk silicon outperforms these low-dimensional materials in terms of normalized sensor response (%). The results suggest that electrostatic charge confinement can be an effective route toward achieving high sensitivity with potential advantages over structural charge confinement.

**CONCLUSION**

To summarize, we have demonstrated chemical-sensitive field-effect transistors on bulk silicon, with an electrically floating ultrathin Ni₀.₃ nmPd₁ nm sensing layer for H₂ gas sensing. Through device modeling and simulations, we have shown that by applying different V_SUB the sensitivity of the CS-FET can be tuned electrically. We have corroborated this by measuring the H₂ sensor response of fabricated Ni–Pd CS-FETs, which results in improved sensor linearity and recovery times. Moreover, this platform exhibits minimal sensor hysteresis and long-term drift. The results presented in this work build a compelling case for bulk silicon CS-FETs from both performance and manufacturability perspectives. This platform provides opportunities in a wide variety of applications such as industrial safety, environmental air quality monitoring, wireless sensor networks, and consumer electronics.

**METHODS**

**CS-FET Device Modeling and Simulation.** CS-FET device simulations in Figure 2 and Figure 4 were carried using Synopsys TCAD (Version M-2016.12). Carrier transport in devices is handled by self-consistently solving Poisson's continuity equation with the drift-diffusion model. The Philips unified model is used for calculating mobility in the devices. Quantum confinement effects upon nanoscale devices are taken into consideration using the density-gradient-based quantization model. The Slotboom and Graaff band-
process (Fuji and wet etching the isolation oxide (in 5:1 buffered hydrofluoric acid) bath at 120 °C for 1 min). Following this, an approximately 300 nm thick phosphosilicate glass (PSG) layer was deposited at 450 °C for 5 min. Following this, an ultrathin Ni sensing layer was deposited in the source and drain contact regions, using thermal evaporation and lift-off in acetone. To achieve ohmic source and drain contacts, nickel silicidation (NiSi) was performed in forming gas using an RTA at 420 °C for 5 min. After this, 50 nm of nickel was then deposited in the source and drain contact regions, using thermal evaporation and lift-off in acetone. To achieve ohmic source and drain contacts, nickel silicidation (NiSi) was performed in forming gas using an RTA at 420 °C for 5 min. Following this, an ultrathin Ni−Pd sensing layer was deposited by sequentially evaporating 1 nm Pd (using e-beam) and then 0.3 nm Ni (using thermal), without any vacuum break. Finally, the sensing layer was annealed in N2 at 150 °C for 1 h postdeposition, which completed the sensor fabrication process.

**CS-FET Fabrication Process.** CS-FET gas sensors were fabricated on prime grade silicon (100) wafers with sheet resistivity in the range of 10−20 ohm-cm. A schematic representing the fabrication process is depicted in Supporting Information S1. Before processing, all wafers were cleaned in a standard piranha (1:4, hydrogen peroxide/sulfuric acid) bath at 120 °C and native oxide was removed using a 10 s dip in 1:10 buffered hydrofluoric acid. First, a 350 nm silicon dioxide was thermally oxidized at 1000 °C, at atmospheric pressure for 55 min. Oxide thickness was verified using fixed angle (70°) ellipsometry. Next, source and drain doping regions in silicon were defined using a standard i-line photolithography process (Fujifilm, photosist: OiR 906-12, developer: OPD-4262) and wet etching the isolation oxide (in 5.1 buffered hydrofluoric acid for 5 min). Following this, an approximately 300 nm thick phosphosilicate glass (PSG) layer was deposited at 450 °C using low-pressure chemical vapor deposition (LPCVD). To complete the formation of n-type doped regions, phosphorus drive-in and activation was performed in the silicon source and drain by rapid thermal annealing (RTA) at 1050 °C for 30 s in N2. The PSG layer was then removed in a 1:10 hydrofluoric acid bath for 1 min. This process step involves overetching that can lead to some loss in field oxide from the original 350 nm, but is inconsequential to the overall device isolation. The “gate” or sensing layer region was patterned next and etched in 5:1 buffered hydrofluoric acid for 4 min. To define source and drain contacts, a separate source−drain metallization mask was used, which underlaps the doped source and drain regions by 11 μm. After this, 50 nm of nickel was then deposited in the source and drain contact regions, using thermal evaporation and lift-off in acetone. To achieve ohmic source and drain contacts, nickel silicidation (NiSi) was performed in forming gas using an RTA at 420 °C for 5 min. Following this, an ultrathin Ni−Pd sensing layer was deposited by sequentially evaporating 1 nm Pd (using e-beam) and then 0.3 nm Ni (using thermal), without any vacuum break. Finally, the sensing layer was annealed in N2 at 150 °C for 1 h postdeposition, which completed the sensor fabrication process.

**Sensor Measurement Apparatus.** All gas-sensing experiments described in this paper were done in a walk-in fume hood. CS-FET device chips were wire bonded to a 28-pin J-bend leaded chip carrier. A small-volume (~0.83 cm³) 3D printed housing (made of polyactic acid) consisting of a 1/4 in. gas inlet was used to cover the chip carrier. Pure dry air was used as diluent gas and was procured from Praxair Technology Inc. For H2 sensing experiments, 5% H2 in N2 (Praxair) was used as source. Ultra-high-purity H2 (Praxair) was used for the experiment in Supporting Information S2. House-compressed dry air was used for week-long extended measurements (in Figure 5c). Typical gas flow rates were from 1 to 100 sccm, and diluent (air) flow rate was approximately 1000 sccm. Ambient temperature and humidity were monitored by commercial sensors purchased from Sensirion AG (models SHT2x and SHT3x). Gas delivery was controlled by mass flow controllers (Alicat Scientific Inc.). CS-FET sensors were biased using a Keithley 428 current preamplifier, and the current signals were acquired using a LabVIEW-controlled data acquisition unit (National Instruments, NI USB-6259). A Keysight 4155C semiconductor parameter analyzer was used for extended length ambient drift measurements in Figure 5b. Measurements in Supporting Information S4 were carried out using a different electronic readout setup and data acquisition board (National Instruments, NI-USB 6218).
ASSOCIATED CONTENT

1 Supporting Information
The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsnano.8b00580.

 Bulk CS-FET fabrication process; H₂ sensor response characteristics of bulk CS-FETs with forming gas annealed Ni–Pd sensing layer; H₂ sensor response characteristics of bare (unfunctionalized) bulk CS-FETs; analysis of H₂ surface concentration, desorption rates, and correlation to sensor recovery at different V_{sub}; ambient drift characteristics of bare (unfunctionalized) bulk CS-FETs; references (PDF)

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H. M. Fahad and N. Gupta contributed equally to this work.

Author Contributions
H.M.F. and N.G. contributed equally to this work. H.M.F. led the project and fabricated the CS-FET sensors as well as device modeling and simulation. N.G. and H.M.F. carried out the measurements and analysis. N.G. and R.H. contributed to measurement setup and programming. S.B.D. contributed to device analysis. A.J. supervised the project. All authors discussed the results and wrote the paper.

Notes
The authors declare the following competing financial interest(s): H.M.F. and A.J. declare competing financial interests in equity on shares of Serinus Labs, Inc.

ACKNOWLEDGMENTS

A.J. acknowledges the Bakar Fellows Program in funding this research work. H.M.F. and N.G. thank H. Kim, D.-H. Lien, A.B. Sachid, and M. Amani for fruitful discussions.

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